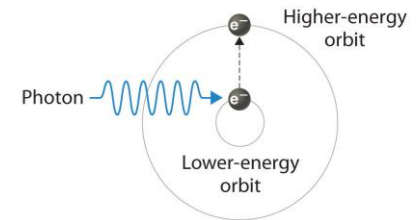
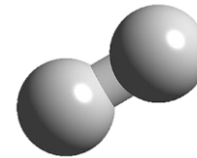




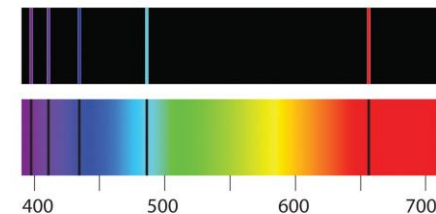
UNITED COLOURS OF HYDROGEN

What is hydrogen?

- Hydrogen is the first element of the periodic table and most abundant element of the universe (~75%).
- At ambient condition hydrogen is a **colourless**, odourless and tasteless bi-atomic gas (H₂).



(a) Electronic absorption transition



(b) H₂ emission spectrum (top), H₂ absorption spectrum (bottom)

Fuel	State @ ambient temp. and pres.	HHV (MJ/kg)	LHV (MJ/kg)	HHV (MJ/l*)	LHV (MJ/l*)
Hydrogen	gas	141,9	119,9	2,3	1,9
Methane	gas	55,5	50,0	7,2	6,5
Gasoline	liquid	47,5	44,5	32,3	30,3
Diesel	liquid	44,8	42,5	38,1	36,1
Methanol	liquid	20,0	18,1	15,8	14,3

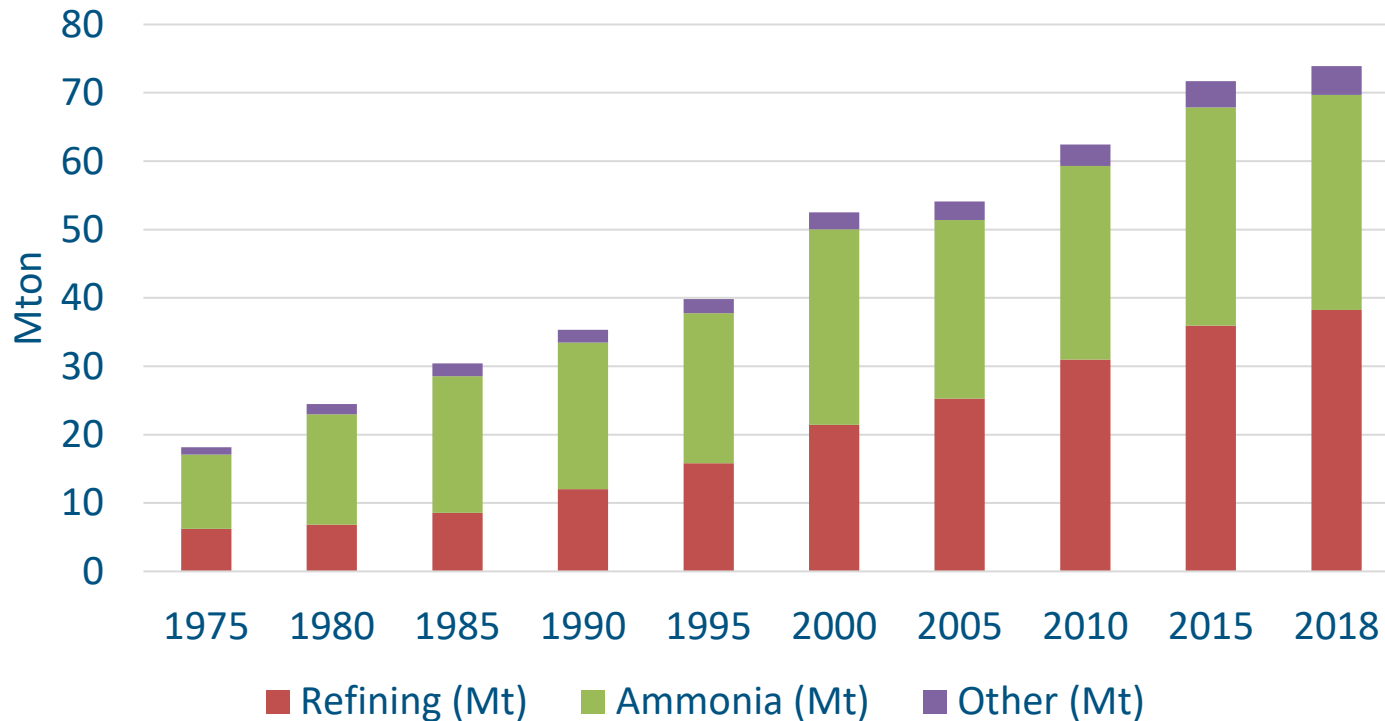
* @25°C & 200 bar for gases

1	2
1 H Idrogeno	
3 Li Litio	4 Be Berillio
11 Na Sodio	12 Mg Magne...
19 K Potassio	20 Ca Calcio
37 Rb Rubidio	38 Sr Stronzio
55 Cs Cesio	56 Ba Bario
87 Fr Francio	88 Ra Radio

What is hydrogen being used for today?

Around **74 million tons/year** worldwide H₂ production (2018)¹.

Global pure hydrogen demand



Future use case as energy carrier → transportation sector decarbonization

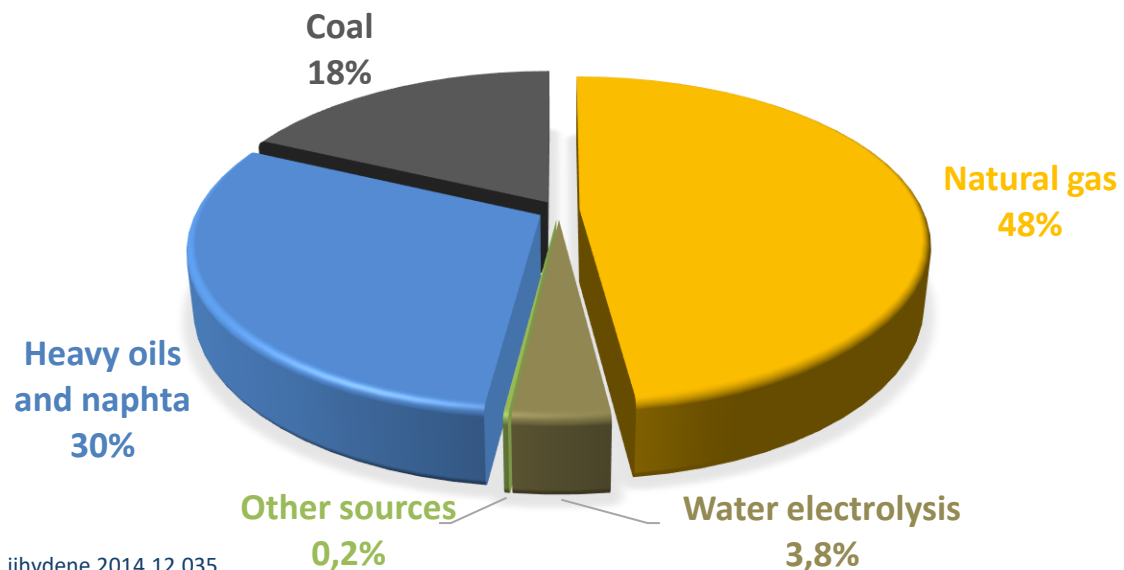
¹ <https://www.iea.org/reports/the-future-of-hydrogen>

How is hydrogen produced today?

Unlike fossil fuels, hydrogen is not readily available in nature. It can be however produced from any primary energy source, for e.g.:

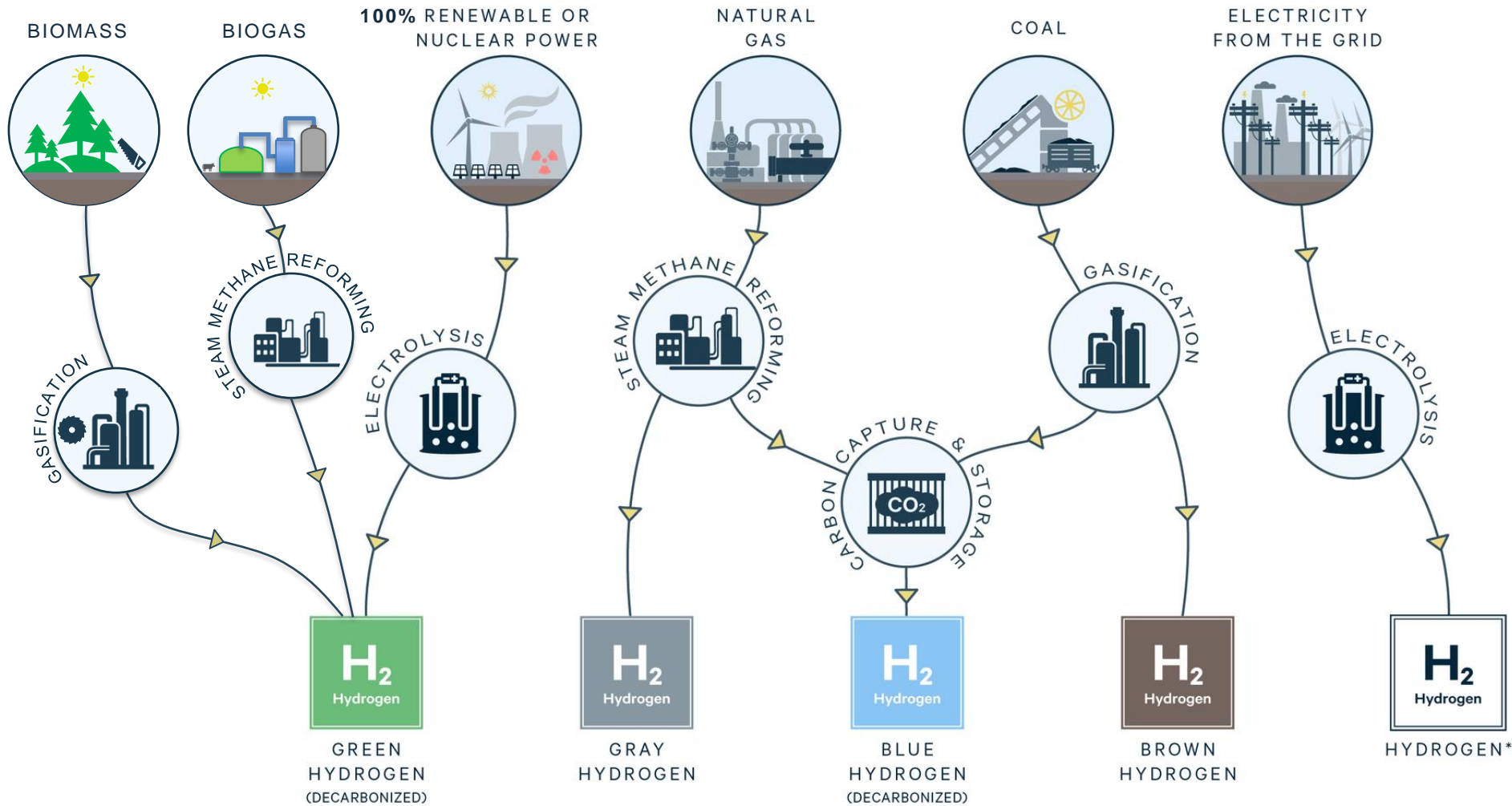
- **Steam reforming + WGS method:** $C_nH_m + 2nH_2O \xrightarrow{\text{heat}} nCO_2 + \left(2n + \frac{1}{2}m\right) H_2$
- **Partial oxidation + WGS method:** $C_nH_m + nO_2 \rightarrow nCO_2 + \frac{1}{2}mH_2$
- **Water electrolysis:** $H_2O \xrightarrow{e^-} H_2 + \frac{1}{2}O_2$

Hydrogen production by source²

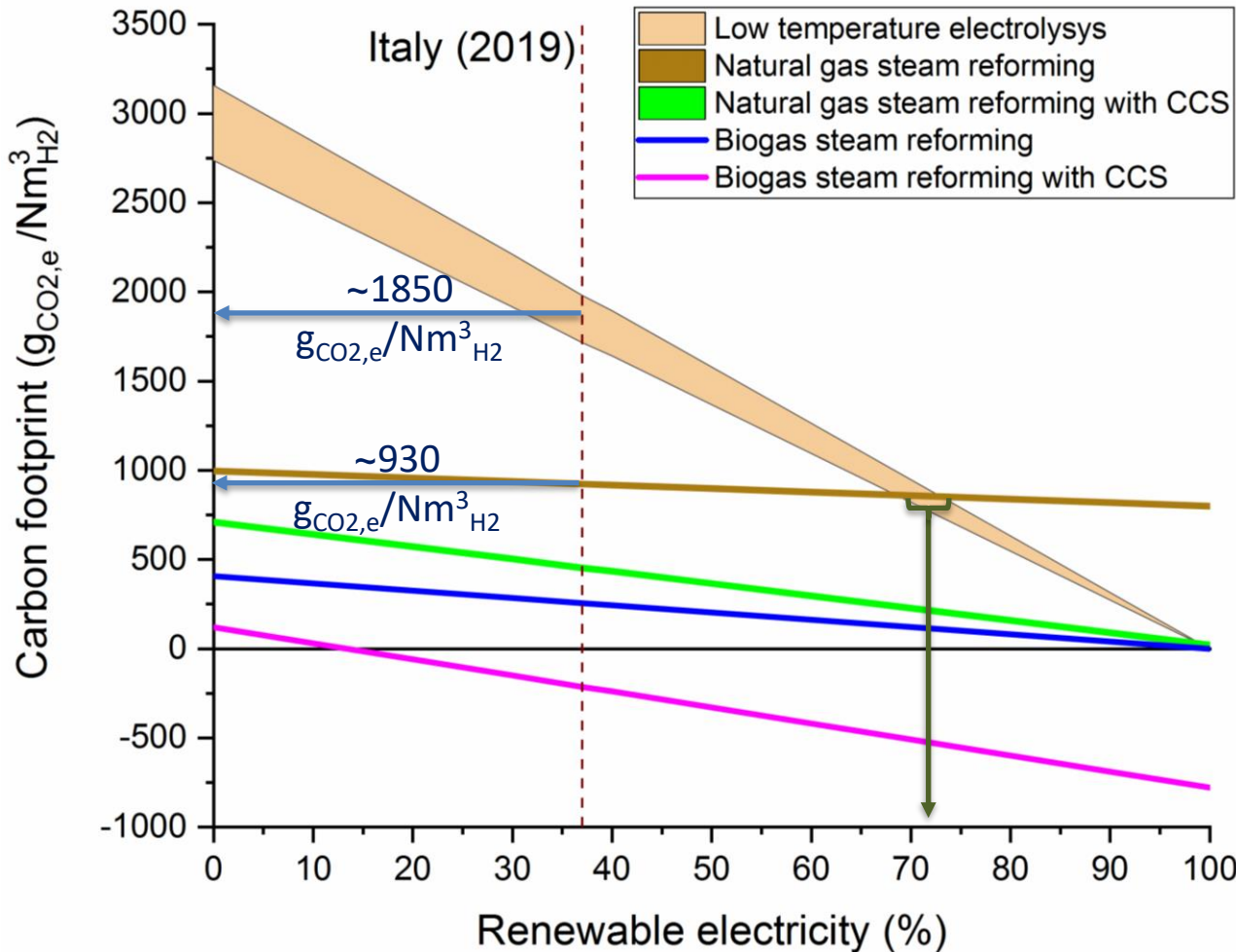


² Dincer et al. (2014), International Journal of Hydrogen Energy; DOI: 10.1016/j.ijhydene.2014.12.035

Why are we talking about colours of hydrogen?



Adapted from: <https://www.rff.org/publications/issue-briefs/investment-tax-credits-hydrogen-storage/>



How is the situation today?

Gray hydrogen is the less impactful hydrogen production method (0,93 kg_{CO_{2,e}}/Nm³ H₂ vs 1,85 kg_{CO_{2,e}}/Nm³ H₂)

When will Electrolysis be competitive with Gray Hydrogen in terms of emissions?

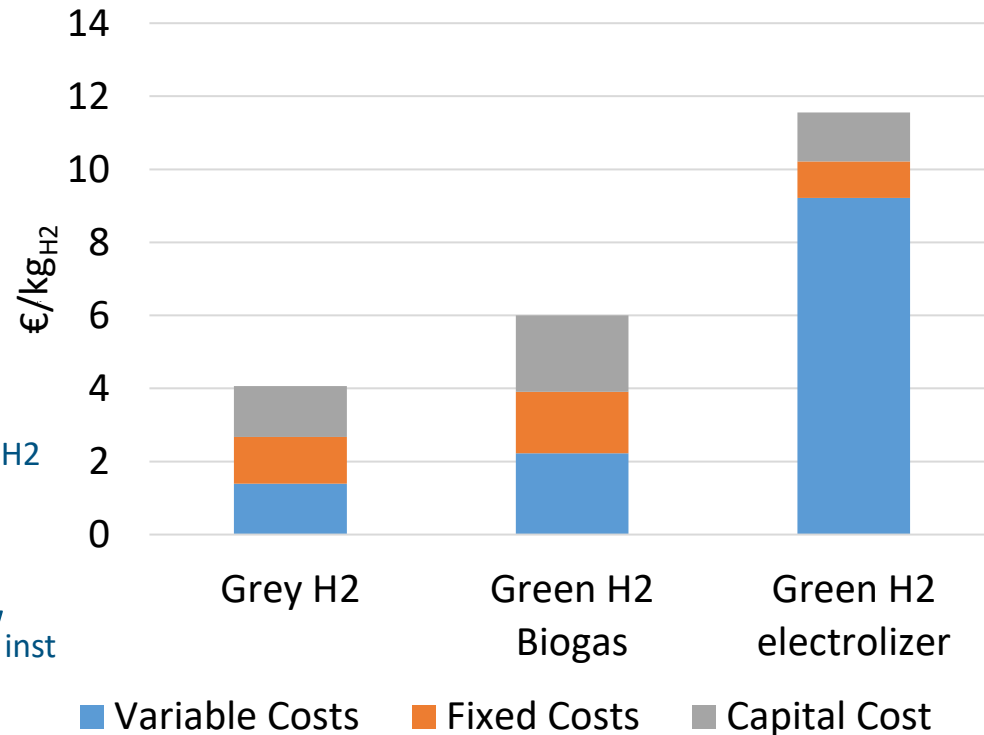
Over 70% of renewable electricity in the energy mix.

But how much does it cost?

Main assumptions:

- **Plant capacity:** 1 M Nm³/y
- **Up-time:** 8300 h/y
- **Electricity cost:** 0,14 €/kWh
- **Natural gas cost:** 0,18 €/Sm³
- **Electrolyser efficiency:** 5,4 kWh/Nm³_{H2}
- **Depreciation time:** 10 years
- **Specific electrolyser cost:** 1200 €/kW_{inst}

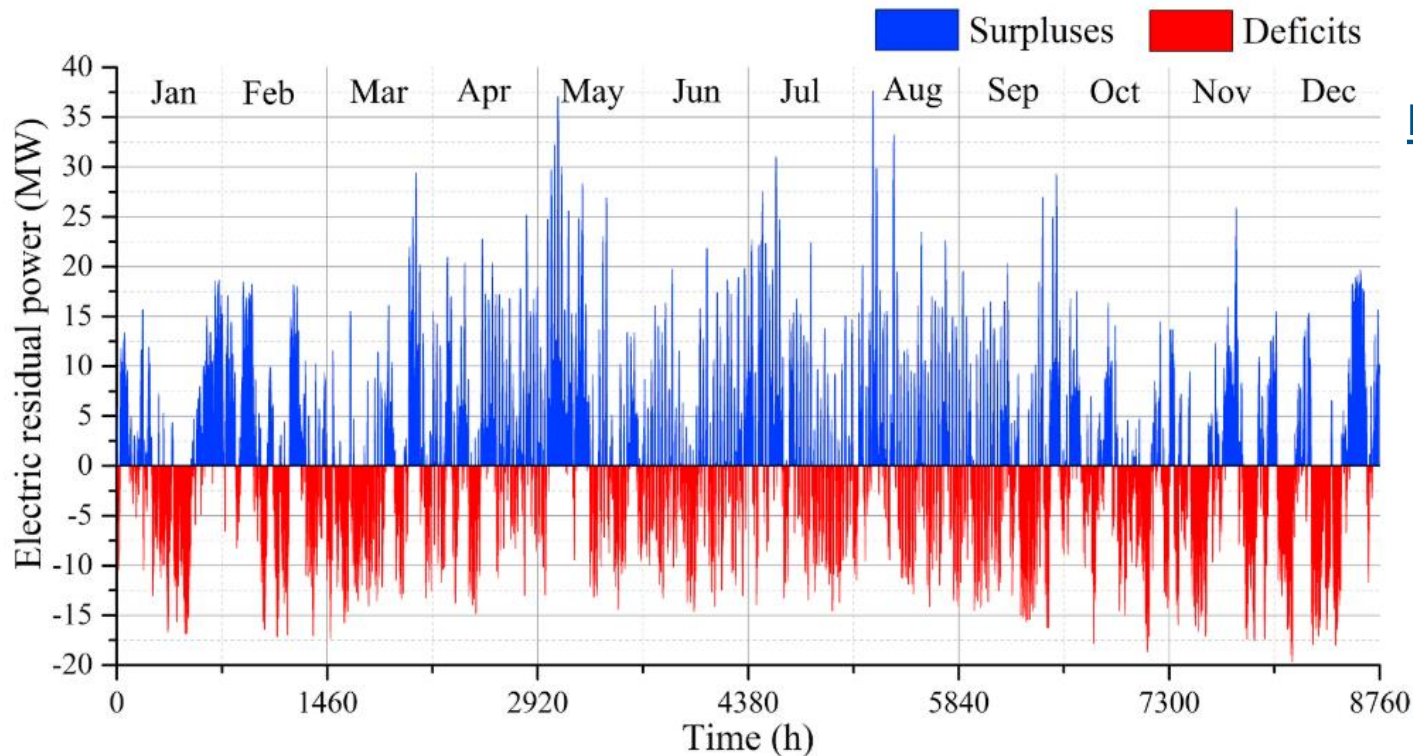
Total cost of ownership (TCO)



Optimistic → Current prices >1800 €/kW_{inst}

Electricity production residuals in a high renewables production scenario (2050 scenario³)

$$\text{Residual} = \text{Production} - \text{Consumption}$$



Electr. Surplus periods

Current Italy:

~1000 h/y

Future scenario:

~3500 h/y

³ Salomone et al. (2018), Chemical Engineering Journal; DOI: 10.1016/j.cej.2018.10.170

How much does it cost?

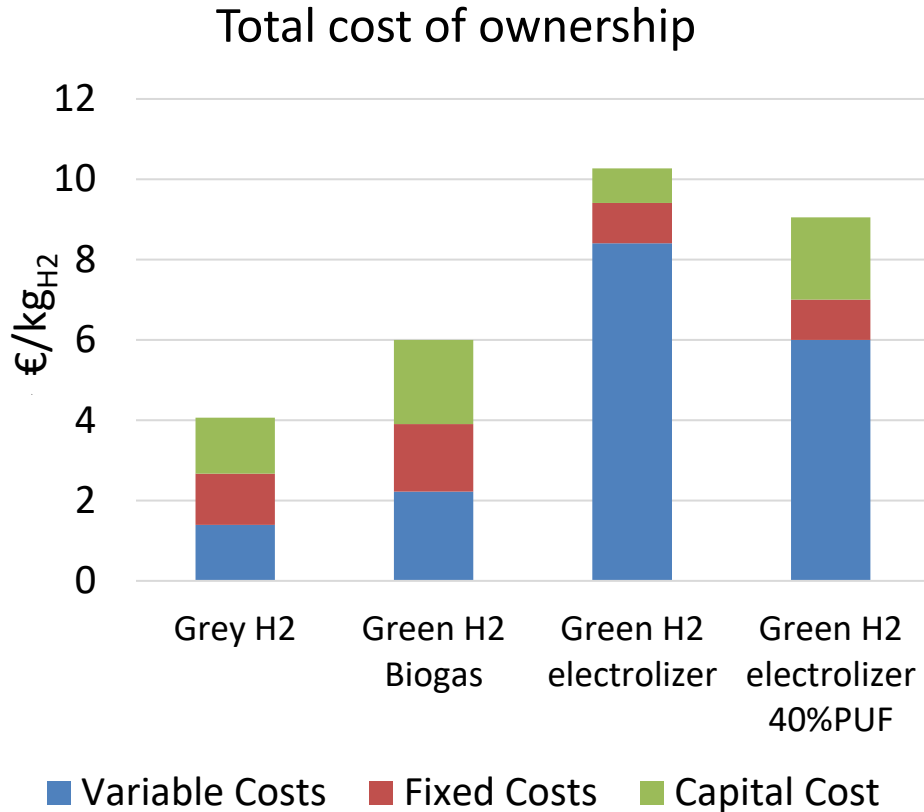
www.hysytech.com

Main assumptions electrolysis only surplus:

- **Plant capacity:** 1 M Nm³/y
- **Up-time:** 3500 h/y
- **Electricity cost:** 0,10 €/kWh
- **Electrolyser efficiency:** 5,4 kWh/Nm³_{H2}
- **Depreciation time:** 10 years
- **Specific electrolyser cost:** 1200 €/kW_{inst}

Electrolysis installed capacity increase:
650 kW → 1500 kW

Optimistic → Current prices >1800 €/kW_{inst}

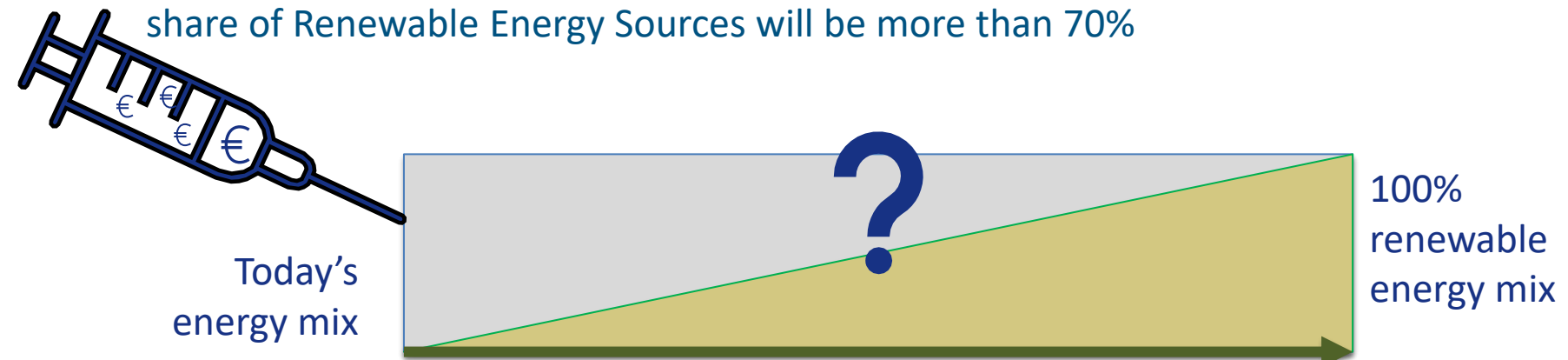


- The target is to install **40 GW** of electrolyzers
- The subsidy program to allow the installation is the biggest plan ever seen to help a technology development
- The electrolyzers have to be fed by 100% Renewable electricity (otherwise a significant increase of CO₂ emission will be obtained)
- Assuming that the electrolyzers will operate 8400 hours/year (requiring huge installation of renewable sources plus huge storages for electricity) the overall production will be around 5,5 Million tons/year, which is the **7% only** of the actual Hydrogen industrial demand
- The energy transition is going to be a pretty **long way**....

- The experience developed already 20 years ago is clearly stating that a new technology can penetrate the market and diffuse only if it's both **economically competitive** and **beneficial for the environment**
- Electrolysers today need an **enormous amount of subsidies** for the **CAPEX** and even for the **OPEX**
- Electrolyser indirect emissions today can be **two times worse for the environment** compared to the Hydrogen obtained from fossil Natural Gas without any CO2 capture.
- The **very first priority** for the environment is not the electrolyser, but is the **increase of the Renewable Energy Sources**.

What can we do in the meantime?

- The demand for hydrogen in the short to medium term is expected to grow, if cheap hydrogen will be available
- **Gray hydrogen** is today and in the mid term the **most environmentally friendly** and **cost efficient** hydrogen production technology if compared to electrolysis fed with electricity from the grid
- Using **biogas and biomass gasification** has the biggest potential in the short to mid term to produce **green hydrogen**
- Retrofit grey hydrogen facilities with CCUS to produce blue hydrogen, when the share of Renewable Energy Sources will be more than 70%



What can we do in the meantime?

www.hysytech.com



H2 Genio

Hydrogen generation directly on site, simply starting from Natural Gas, from small flow rates and up to industrial sizes.

Plants in Italy



Biomethane

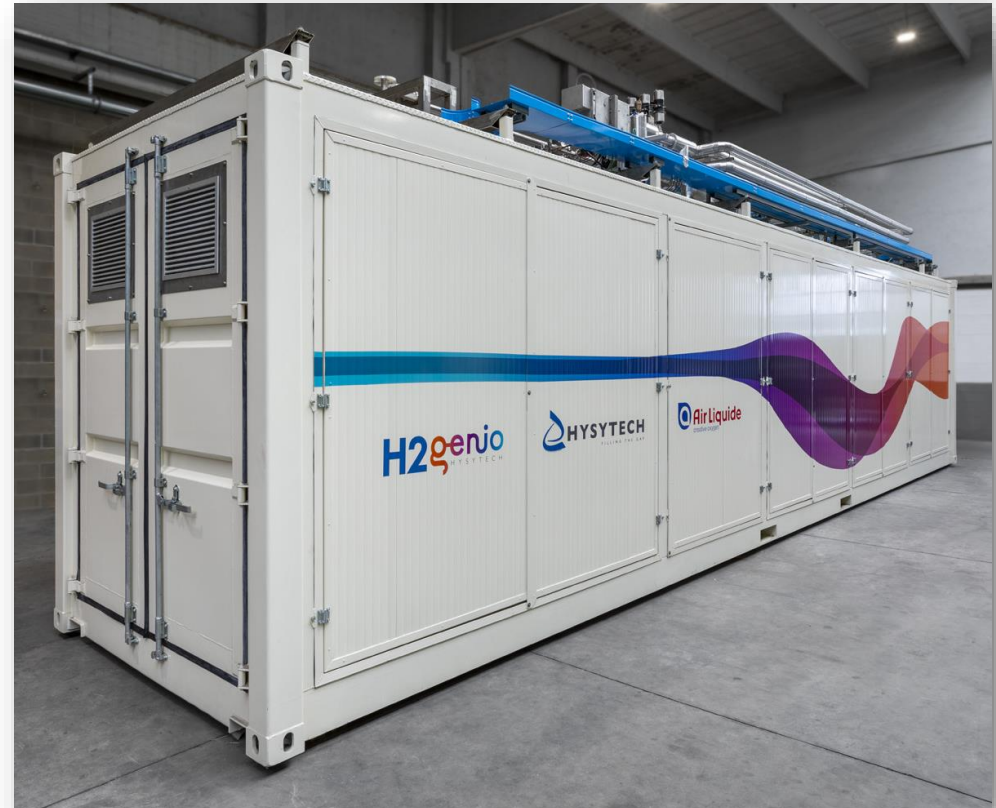
Biomethane from raw biogas.

Plants in Italy from MSW and Biomass

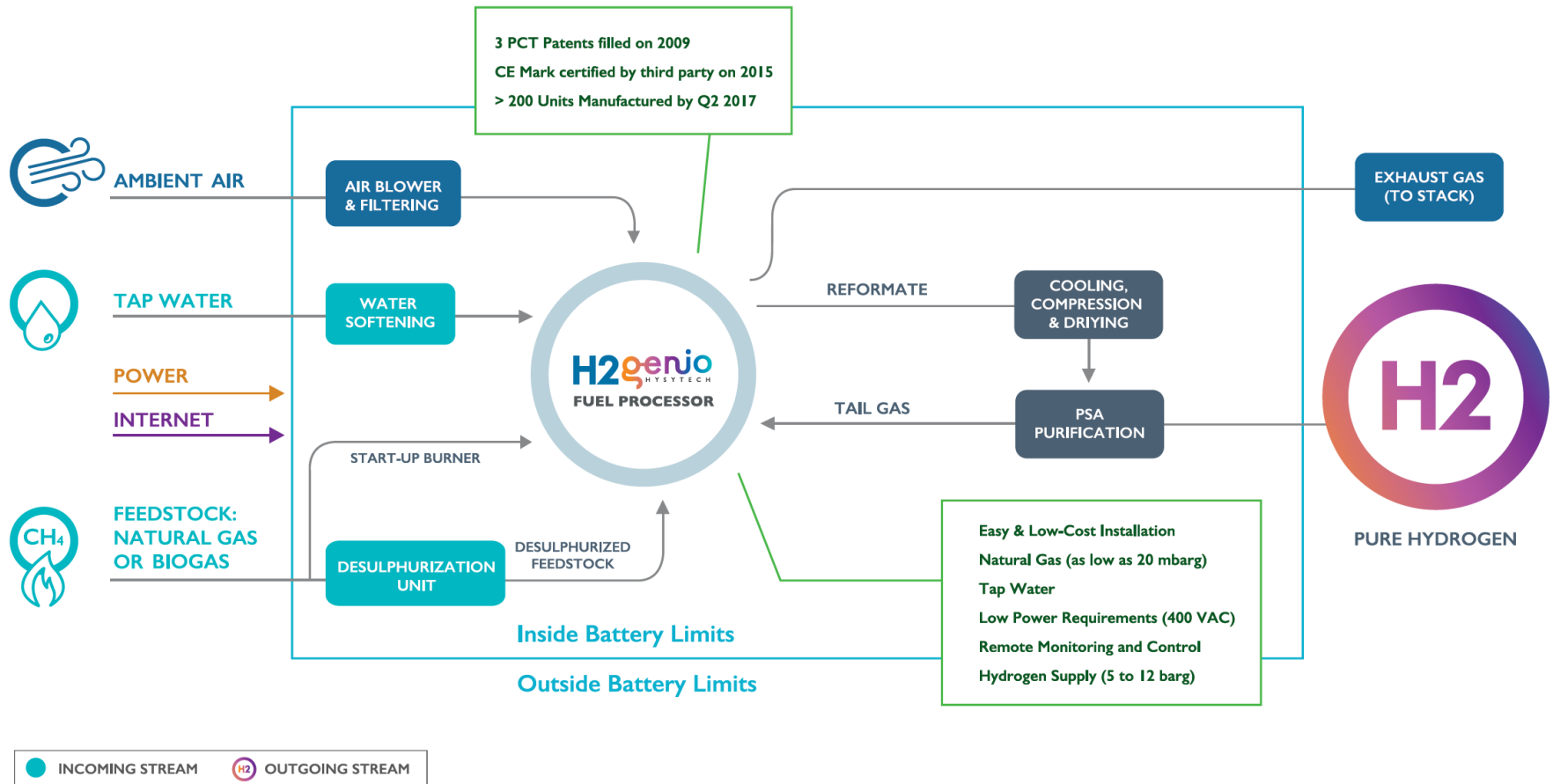
• Hight Purity Hydrogen for Steel Industry (120 Nm³/h)

Features:

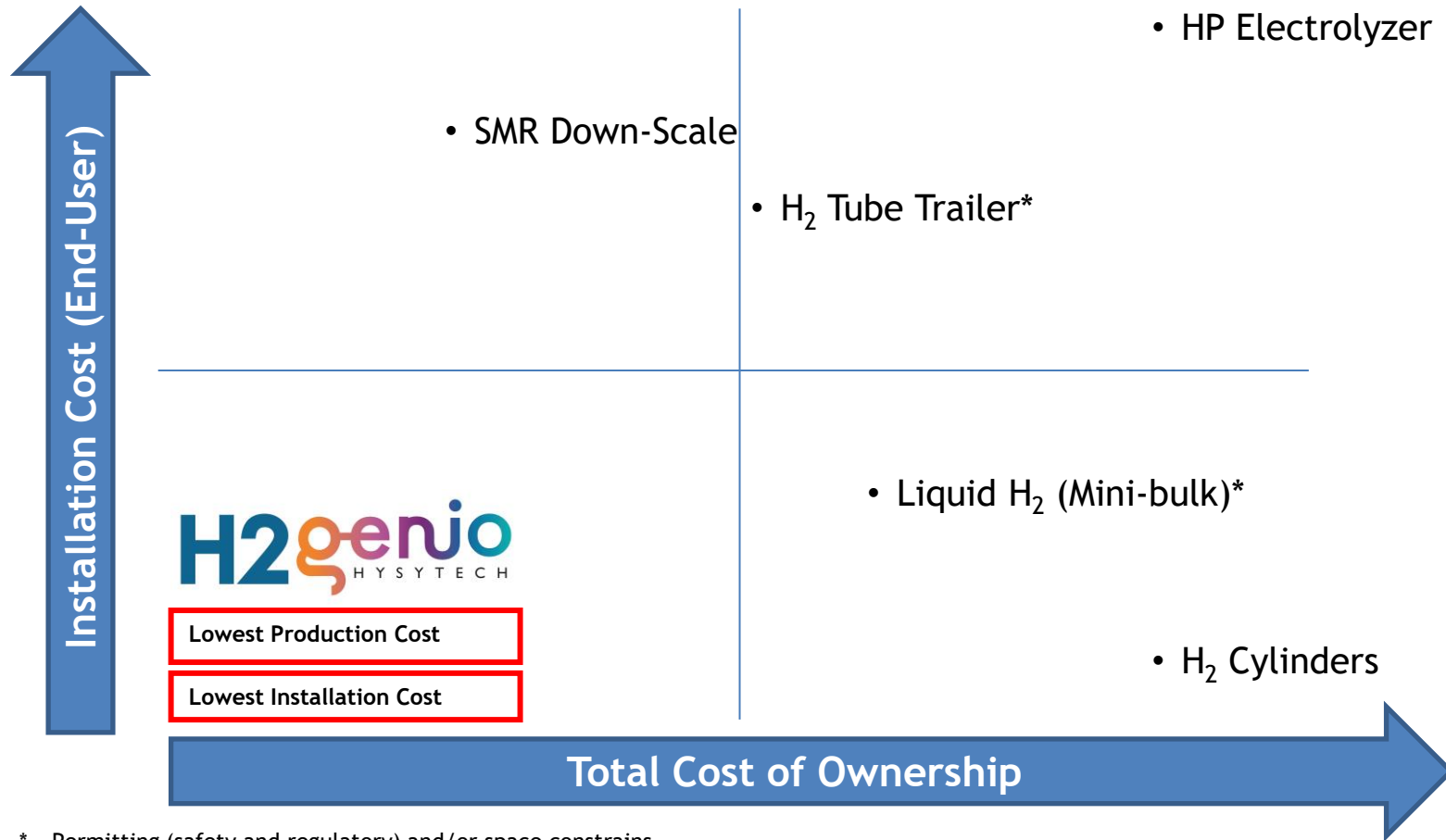
- High Purity Hydrogen >99,995%vol
- Capacity up to 120 Nm³/h
- Power Consumption <40 kW
- Delivery Pressure 11 barg
- Natural Gas feed:
 - Pressure <0,350 mbarg
 - Odorized Gas
- Fully automatic & unmanned



Turn-key Product Battery Limits



• Known Competitors & Alternatives



* Permitting (safety and regulatory) and/or space constrains often make this kind of delivery not possible



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